

Supplementary material

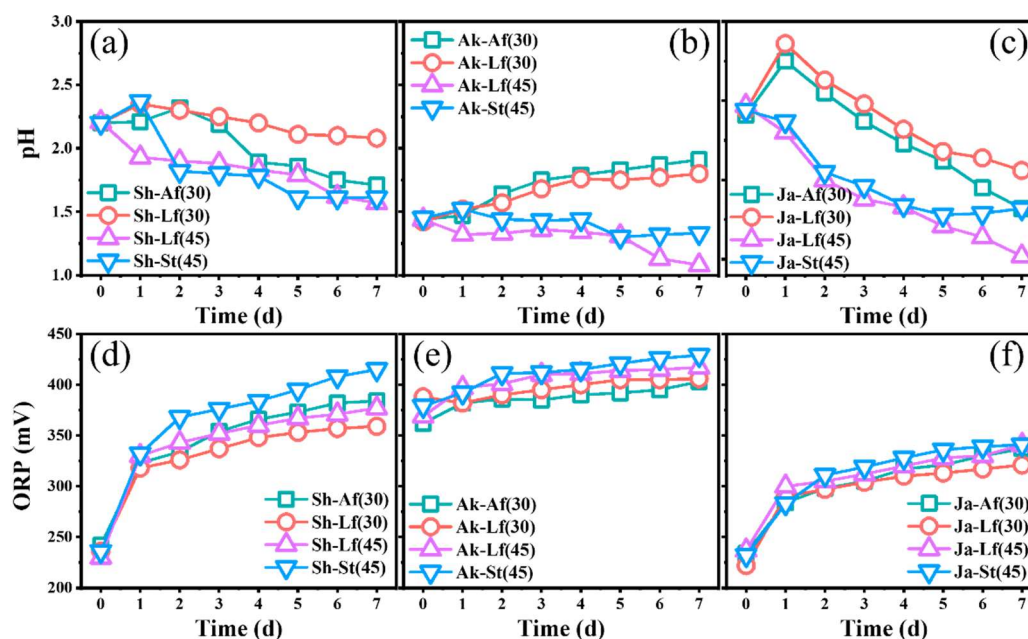


Figure S1 pH (a, b, c) and ORP (d, e, f) variation during iron minerals formation process

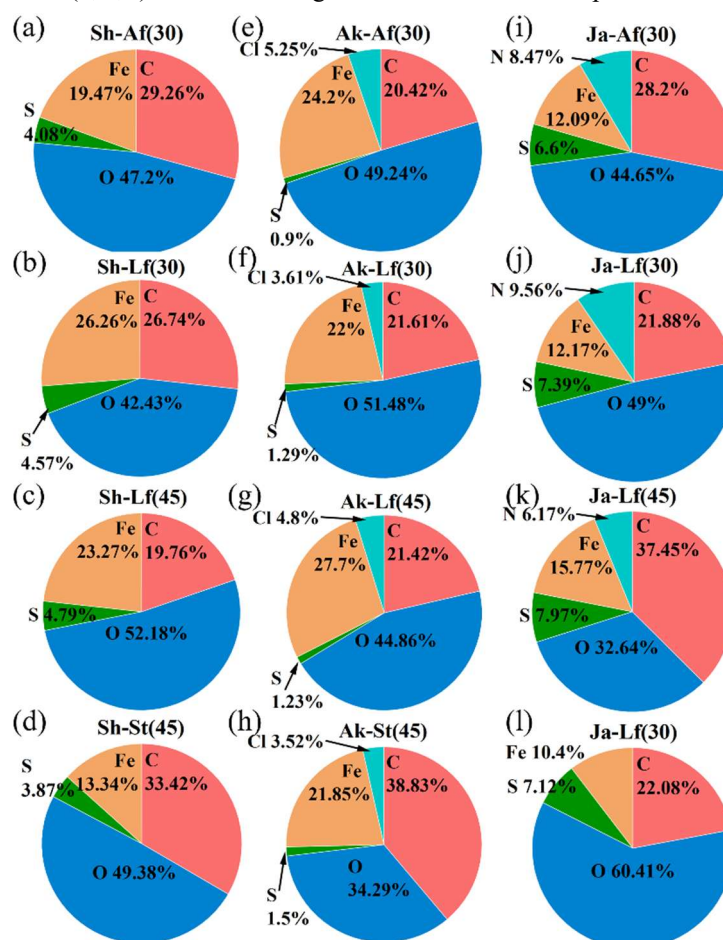


Figure S2 The EDS datas of the precipitate. Schwertmannite (a–d), Akaganéite (e–h) and Ammoniojarosite (i – l) biosynthesized by *A. ferrooxidans* (30 °C), *L. ferrooxidans* (30 °C), *L. ferriphilum* YSK (45 °C), and *S. thermosulfidooxidans* (45 °C) respectively

Table S1 Parameters of Schwertmannite, Akaganéite and Ammoniojarosite

Bacteria	T(°C)	Q (g)	Fe (%)	S (%)	Cl (%)	O (%)	N (%)	(Fe/S, Cl, N) mole ratio	Formula	XRD peak	XS (nm)
<i>A. ferrooxidans</i>	30	3.28	46.77	5.62	/	32.48	/	4.755	$\text{Fe}_8\text{O}_8(\text{OH})_{4.64}(\text{SO}_4)_{1.68}$	0	non-crystal body
<i>L. ferrooxidans</i>	30	1.08	56.12	5.61	/	25.98	/	5.716	$\text{Fe}_8\text{O}_8(\text{OH})_{5.20}(\text{SO}_4)_{1.40}$	0	non-crystal body
<i>L. ferrophilum</i> YSK	45	2.12	51.46	6.08	/	33.06	/	4.836	$\text{Fe}_8\text{O}_8(\text{OH})_{4.69}(\text{SO}_4)_{1.65}$	0	non-crystal body
<i>S. thermosus</i>	45	3.56	36.15	6.02	/	38.35	/	3.431	$\text{Fe}_8\text{O}_8(\text{OH})_{3.33}(\text{SO}_4)_{2.33}$	5	non-crystal body
<i>A. ferrooxidans</i>	30	0.47	51.99	/	7.16	30.31	/	4.603	$\text{Fe}_8\text{O}_8(\text{OH})_{6.26}\text{Cl}_{1.74}$	6	381
<i>L. ferrooxidans</i>	30	0.44	49.52	/	5.16	33.19	/	6.084	$\text{Fe}_8\text{O}_8(\text{OH})_{6.68}\text{Cl}_{1.31}$	2	179
<i>L. ferrophilum</i> YSK	45	0.78	56.63	/	6.23	26.28	/	5.762	$\text{Fe}_8\text{O}_8(\text{OH})_{6.61}\text{Cl}_{1.39}$	4	358
<i>S. thermosus</i>	45	1.1	50.67	/	5.19	22.78	/	6.189	$\text{Fe}_8\text{O}_8(\text{OH})_{6.70}\text{Cl}_{1.29}$	2	167
<i>A. ferrooxidans</i>	30	3.24	32.8	10.27	/	34.71	5.76	1.826, 1.423	$\text{Fe}_8\text{O}_8(\text{OH})_{4.85}(\text{NH}_4)_{5.62}(\text{SO}_4)_{4.38}$	13	390
<i>L. ferrooxidans</i>	30	1.85	32.4	11.3	/	37.38	6.39	1.639, 1.268	$\text{Fe}_8\text{O}_8(\text{OH})_{4.55}(\text{NH}_4)_{6.31}(\text{SO}_4)_{4.88}$	16	411
<i>L. ferrophilum</i> YSK	45	4.13	40.12	11.64	/	23.8	3.94	1.970, 2.548	$\text{Fe}_8\text{O}_8(\text{OH})_{3.01}(\text{NH}_4)_{3.14}(\text{SO}_4)_{4.06}$	19	694
<i>S. thermosus</i>	45	4.32	28.46	11.18	/	47.36	3.34	1.455, 2.128	$\text{Fe}_8\text{O}_8(\text{OH})_{0.76}(\text{NH}_4)_{3.76}(\text{SO}_4)_{5.50}$	18	499

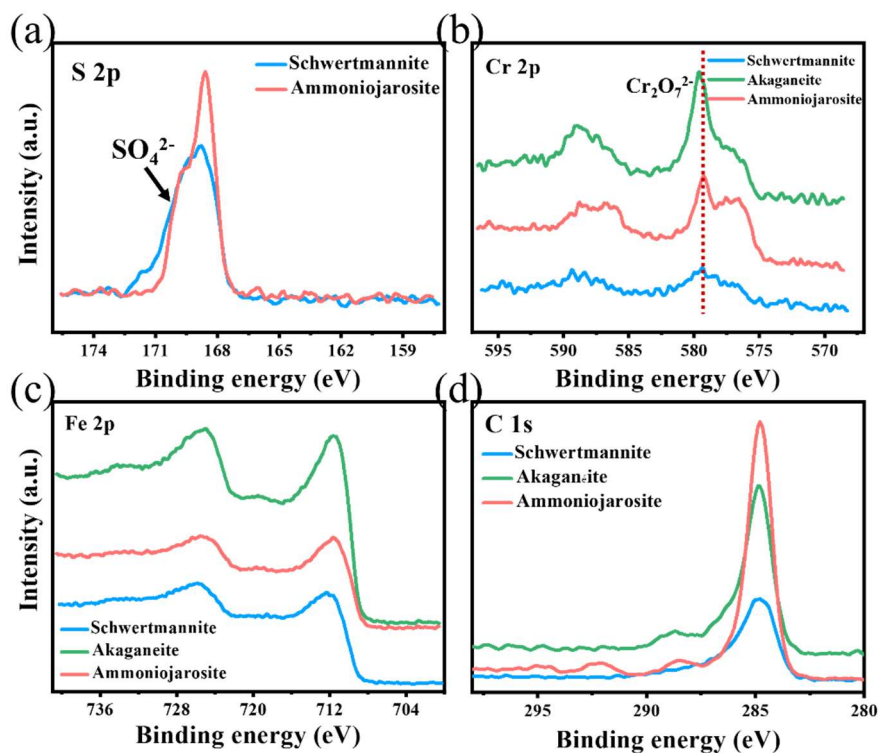


Figure S3 XPS spectra of the precipitates

Table S2 Surface element composition (Atomic %) of the materials after Cr(VI) adsorption

	C	O	Cr	Fe	S	Cl
Schwertmannite	24.54	56.55	0.88	12.58	5.45	/
Akaganéite	24.79	54.19	1.3	16.83	/	2.88
Ammoniojarosite	39.13	43.6	1.35	9.73	4.27	/

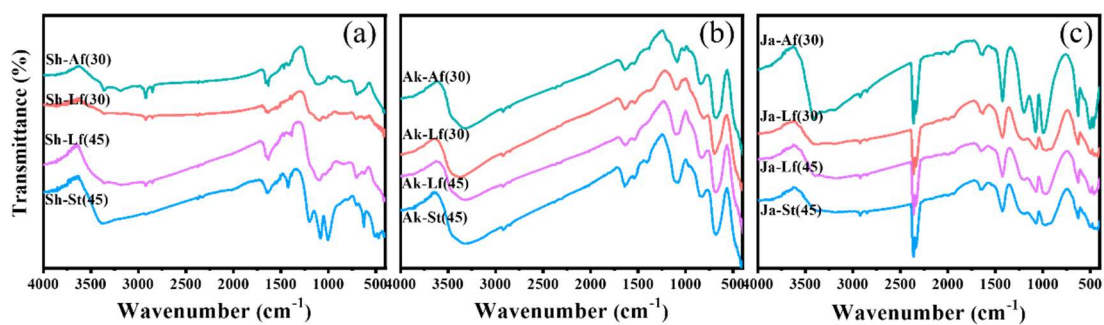


Figure S4 FTIR spectra of the precipitates after Cr(VI) adsorption